Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

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## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.007 \AA$
$R$ factor $=0.073$
$w R$ factor $=0.208$
Data-to-parameter ratio $=13.9$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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## Methyl 3,5-diacetoxy-4-methoxybenzoate

The title compound, $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{O}_{7}$, is derived from methyl gallate, a protective biological antioxidant. It crystallizes with two independent molecules in the asymmetric unit.

## Comment

Methyl gallate (methyl 3,4,5-trihydroxybenzoate) can be isolated from Acer truncatum Bunge and it is considered to be a major pharmacological antioxidant (Whang et al., 2005). It is widely distributed and is used to treat coronary arteriosclerosis, cerebrovascular diseases and angina pectoris (Liu, 2003). In order to investigate a similar biological compound derived from methyl gallate, the title compound, (I), was prepared and its crystal structure is reported here.

(I)

The asymmetric unit consists of two independent molecules of (I) (Fig. 1). In both molecules, the bond lengths and angles are within normal ranges (Allen et al., 1987). The C1-C6 and C14-C19 benzene rings are planar, with r.m.s. deviations of 0.0043 and $0.0085 \AA$, respectively. The two independent molecules in the asymmetric unit are very similar in conformation, but show significant small differences in the magnitudes of their torsion angles. For example, the torsion angles $\mathrm{C} 9-\mathrm{O} 5-$ $\mathrm{C} 2-\mathrm{C} 3\left[-73.6(5)^{\circ}\right]$ and $\mathrm{C} 11-\mathrm{O} 7-\mathrm{C} 6-\mathrm{C} 1\left[-77.0(5)^{\circ}\right]$ are comparable to those for $\mathrm{C} 24-\mathrm{O} 11-\mathrm{C} 19-\mathrm{C} 18\left[88.7\right.$ (5) $\left.{ }^{\circ}\right]$ and C22-O13-C15-C14 [82.9 (5) ${ }^{\circ}$ ].

In the crystal packing, the molecules are linked by several weak intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions, shown as dashed lines in Fig. 2.

## Experimental

All the solvents and reagents were commercial and purified according to standard laboratory techniques. To a solution of methyl 3,4,5triacetoxybenzoate ( $1 \mathrm{mmol}, 310 \mathrm{mg}$ ) in DMF ( 10 ml ) was added $\mathrm{K}_{2} \mathrm{CO}_{3}(3 \mathrm{mmol}, 414 \mathrm{mg})$ and $\mathrm{CH}_{3} \mathrm{I}(2 \mathrm{mmol}, 0.13 \mathrm{ml})$. After stirring for 3 h at 413 K , the inorganic salt in the resulting mixture was removed by filtration, and the filtrate was diluted with EtOAc. The organic phase was washed with distilled water, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated to give 279 mg ( $90 \%$ yield) of methyl 3,5 -diacetoxy-4-

Received 30 May 2006
Accepted 17 November 2006

## organic papers

methoxybenzoate after recrystallization from EtOAc-heptane (3:1) (m.p. 356 K ).

## Crystal data

$\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{O}_{7}$

## $Z=8$

$M_{r}=282.24$
Monoclinic, $P 2_{1} / c$
$a=9.0530(18) \AA$
$b=25.769$ (5) A
$c=11.864$ (2) $\AA$
$\beta=92.57$ (3) ${ }^{\circ}$
$V=2764.9(10) \AA^{3}$

## Data collection

Bruker APEX area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\text {min }}=0.967, T_{\text {max }}=0.984$

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0788 P)^{2}\right. \\
& \quad+0.5721 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.27 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.31 \mathrm{e} \AA^{-3}
\end{aligned}
$$

$w R\left(F^{2}\right)=0.208$
$S=1.05$
5146 reflections
369 parameters
H -atom parameters constrained
$D_{x}=1.356 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\mu=0.11 \mathrm{~mm}^{-1}$
$T=293$ (2) K
Prism, colorless
$0.30 \times 0.20 \times 0.15 \mathrm{~mm}$

5484 measured reflections 5146 independent reflections 2421 reflections with $I>2 \sigma(I)$ $R_{\text {int }}=0.164$ $\theta_{\text {max }}=25.5$

The residual factor for equivalent reflections is large ( $R_{\text {int }}=0.16$ ) because the data are very weak. H atoms were included in calculated positions and refined using the riding-model approximation, with $\mathrm{C}-$ $\mathrm{H}($ aromatic $)=0.93 \AA$ and $\mathrm{C}-\mathrm{H}($ methyl $)=0.96 \AA$, and with $U_{\text {iso }}(\mathrm{H})$ $=1.2 U_{\text {eq }}(\mathrm{C})$ for aromatic and $U_{\text {iso }}(\mathrm{H})=1.5 U_{\text {eq }}(\mathrm{C})$ for methyl H atoms.

Data collection: SMART (Bruker, 2003); cell refinement: SAINT (Bruker, 2003); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXL97.

The support by the Modern Analytical Center at Nanjing University is gratefully acknowledged.

## References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. \& Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.

Bruker (2003). SAINT and SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
Liu, X.-Y. (2003). Chem. Technol. (Chin.), 30, 27-28.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.


Figure 1
The asymmetric unit of the title compound, (I), showing $30 \%$ probability displacement ellipsoids and the atom-numbering scheme.


Figure 2
The crystal packing of (I), viewed along the $a$ axis. Weak intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions are shown as dashed lines.

Sheldrick, G. M. (1997a). SHELXL97 and SHELXS97. University of Göttingen, Germany.
Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
Whang, W.-K., Park, H. S., Ham, I., Oh, M., Namkoong, H., Kim, H. K., Hwang, D. W., Hur, S. Y., Kim, T. E., Park, Y. G., Kim, J. R. \& Kim, J. W. (2005). Exp. Mol. Med. 37, 343-352.


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